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Question: 1.11 - Name the parameters that characterise a unit cell.

Answer: Unit cells are characterize on six parameters - dimensions along three edges and three angles between their edges, i.e. a , b , c which are edges and α , β and γ which are angles between the edges.

Question: 1.12 - Distinguish between (i) Hexagonal and monoclinic unit ...

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10.2. Why is sulphuric acid not used during the reaction of alcohols with KI? To solve the problem, H_2SO_4 is replaced by phosphoric acid (H_3PO_4) which provides HI for the reaction and does not give I_2 as is done by H_2SO_4 . 10.3. Write structures of different dihalogen derivatives of propane. 10.7 ...

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Question 3.11: Suggest a list of metals that are extracted electrolytically.

Question 3.12: Consider the reaction: $\text{Cr}_2\text{O}_7^{2-} + 14\text{H}^+ + 6\text{e}^- \rightarrow 2\text{Cr}^{3+} + 8\text{H}_2\text{O}$ What is the quantity of electricity in coulombs needed to reduce 1 mol of $\text{Cr}_2\text{O}_7^{2-}$?

Question 3.14: Suggest two materials other than hydrogen that can be used as fuels in fuel cells.

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steps and explains the answer, which helps you to understand the pattern of questioning.

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Electrochemistry

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Question 1: How would you determine the standard electrode potential of the system $\text{Mg}^{2+} | \text{Mg}$?. Answer: We know that it is not possible to measure the electrode potential of a half cell. We can only measure the difference between electrode potentials of two half cells, which gives the cell potential of the cell.

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